

Life cycle assessment of ammonia/hydrogen-driven marine propulsion

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Abstract

Marine fuels are the main sources of pollution from shipping industry. Hydrogen and ammonia have been suggested to be alternative fuels for shipping as these two fuels do not emit carbon dioxides in the combustion process. This study employed life cycle assessment method to compare the environmental performance of propulsion systems using hydrogen and ammonia as marine fuels to fossil fuels. 2-stroke and 4-stroke engines of tankers using fossil fuels were chosen as base case scenarios. Alternative scenarios using 'green' and 'blue' hydrogen and ammonia with the support of pilot fuel were then compared to the base case scenarios. While the performance of the coming combustion concepts for hydrogen and ammonia engines are still unknown, preliminary estimations were used in this study. The results showed that hydrogen and ammonia could substantially reduce the global warming potential, compared with the fossil fuel scenarios. Hydrogen and ammonia are also expected to be highly effective in cutting down the particulate matter and the emission of black carbon.

Keywords

LCA, emissions, environmental impacts, hydrogen, ammonia, marine engine, life cycle

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Introduction

With the majority of cargo capacity transported by sea, marine transport plays an important role in international trade.¹ The pollution from shipping is gaining increasing attention, due to the huge amount of emissions exhausted from ship operation.^{2–4} IMO regulated air pollution in Annex VI of the MARPOL convention. For example, Chapter 3 regulations 13–15 limit NO_x, SO_x and VOCs from shipping.⁵ Chapter 4 regulations 19–28 address the greenhouse gas emissions from fuel use, with IMO Lifecycle GHG – carbon intensity guidelines currently under development, and expected to be agreed upon at MEPC 80 in 2023.

Shipping decarbonization is a critical part of meeting the Paris Agreement target: to meet global warming to 1.5°C and to build a zero-emissions planet.⁶ IMO's ambitious GHG target is to reduce the CO₂ emissions per transportation work from the shipping industry by 40% by 2030 and 70% by 2050, in comparison to 2008. The vision and ambitions of IMO within this century is to achieve zero-emissions goal.⁷ The initial IMO strategy is divided into three periods: short-term 2018–2023,

mid-term 2023–2030 and long-term 2030–2100. Considering that the life span of cargo vessels is 25–30 years, the present is the right time to find new emission reduction mechanisms for the shipping industry to achieve the ambitious target. Marine fuels are crucial actors and have important roles in IMO's strategy.

Nowadays, the huge amount of fossil fuels, for example, HFO, MDO, LNG, etc. used in the shipping industry is the main contributor to greenhouse gas emissions to the atmosphere. Some new solutions are suggested to lighten the impacts from shipping on the environment. These include alternative fuels (H₂, NH₃, methanol, etc.), hybrid propulsion systems, carbon capture and storage techniques, renewable energy (from

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wind, solar or wave).^{8–11} The benefits of these solutions need to be further discussed as each solution has both advantages as well as limitations. Among the alternative fuels in shipping, NH_3 and H_2 seem to be competitive candidates as the combustion of these fuels does not emit the carbon emissions.¹² However, from a life cycle perspective, the upstream process (production phase) of these fuels needs careful consideration.

H_2 is the most common chemical element and it is a promising option for green energy in the future.¹³ The main product from H_2 combustion is water, though depending on the process used, by-product of this process are NO_x and the release of unburned H_2 . Currently, the majority of H_2 is also produced by using the steam-methane reforming method with/without CCS technology.¹⁴ H_2 is also produced by using the water separation technique (electrolysis) using electricity and water. Even though renewable energy such as solar and wind energy can be used, H_2 production still has impacts on the environment due to the need for land, material consumption in infrastructure, and the need for large investments. Another barrier of using H_2 as a marine fuel is the storage of H_2 on board. The space required for H_2 storage is larger than the other fuels such as LNG, NH_3 and MDO.¹⁵ If H_2 is stored in liquid form at very low temperature (-253°C) it either has a limited lifetime before the evaporation, or it requires additional energy for keeping in liquid form.¹⁶ H_2 can be stored in pressurized form, but has then a limited energy density (5.15 GJ/m^3 at 800 bar), compared to 8.55 GJ/m^3 in liquid form.¹⁷ Given the higher energy density, liquid hydrogen was assumed to be the most appropriate form of bunkering for long-distance shipping in this work.

H_2 and NH_3 are often categorized using a taxonomy of different colour codes. ‘Grey’ H_2 is produced from natural gas or coal by using steam-methane reforming or gasification processes. If carbon emissions from these processes are captured, the product will be ‘blue’ H_2 . The carbon capture rate by when producing ‘blue’ H_2 is normally from 85 to 95%.¹⁸ ‘Green’ H_2 is produced by using electrolysis to separate H_2 and oxygen from water using electricity from renewable sources (e.g. wind, solar, wave energy, etc.). Nowadays, the usage of ‘green’ H_2 is still limited.¹⁹

NH_3 is produced by using a chemical synthesis technique, the Haber-Bosch process (Figure 1). The taxonomy of colour codes for NH_3 depends on the type of

H_2 used in the synthesis process. Compared with H_2 , NH_3 requires less space for on-board energy storage. Moreover, NH_3 is also considered as a balanced solution with high volumetric energy and more practical storage characteristics (Table 1).

The LCA is considered a comprehensive method to evaluate the environmental performance of a product or service.²⁰ This method has been applied in maritime industry recently. Bengtsson et al.²¹ compared the LCA of LNG and the fossil fuels and indicated the need of LCA when estimating the environmental impacts of marine fuels. The LCA of alternative fuels were studied in.^{5,22,23} Recently, under the HyMethShip concept, the LCA method was also used to evaluate the environmental the performance of propulsion systems using onboard carbon capture technique.^{24,25} The framework of LCA of marine engines was established^{26,27} then were used to investigate the life cycle performance of a tugboat.²⁸ The reader should refer to a comprehensive review²⁹ for the application of LCA in maritime sector.

Some research papers also studied the environmental impacts of H_2 and NH_3 . For instance, the comparison of blue H_2 and fossil fuel was presented by Howarth et al.,¹⁴ showing that the use of blue H_2 may lead to higher environmental impacts than fossil fuel. Perčić et al.³⁰ also indicated that the fuel cell system could reduce up to 84% of greenhouse gases when green NH_3 is used. Bicer et al.³¹ discussed the possibility of using H_2 in shipping transportation from a life cycle perspective. Hwang et al.³² compared the LCA of natural gas and marine gas oil for the ship operating in Korea.

By using LCA and life-cycle cost assessment, Perčić et al.^{33,34} and Fan et al.³⁵ showed that a battery-powered vessel has lower environmental footprint and low cost than a diesel engine-powered vessel. For fishing trawlers, Koričan et al.³⁶ indicated that soybean–biodiesel–diesel blend and LNG could reduce greenhouse gases and have positive impact on the life cycle cost.

To clarify the benefit of H_2 and NH_3 in shipping industry, this study employed the LCA method to investigate the environmental impacts of marine engines using H_2 and NH_3 as marine fuels (comprising both ‘blue’ and ‘green’ H_2 and NH_3). The results were compared to the environmental impacts of marine engines using fossil fuels.

The study is structured as follows. After the introductory part, the LCA method applied in this study is

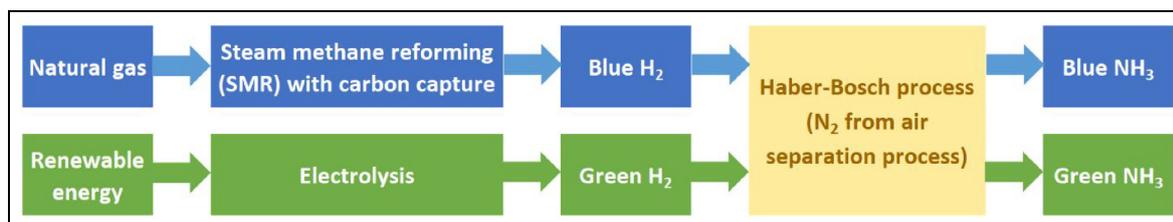
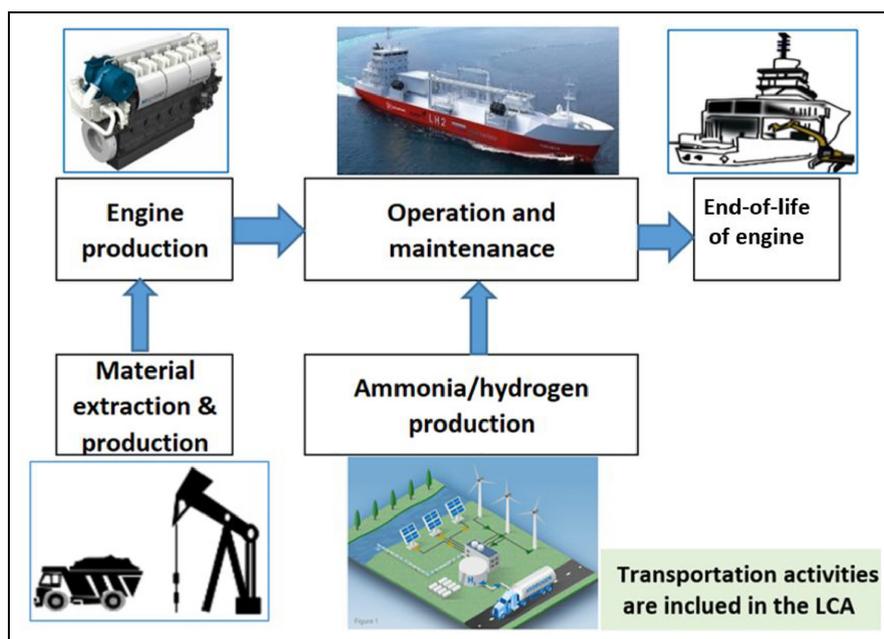


Figure 1. NH_3 and H_2 production processes.

Table 1. Comparison of fuel properties.¹⁵

Fuel type	Energy content [MJ/kg]	Volumetric energy density [GJ/m ³]	Storage pressure [bar]	Storage temperature [°C]
Marine Gas Oil	42.7	36.6	1	20
Liquid Methane	50.0	23.4	1	−162
Ethanol	26.7	21.1	1	20
Methanol	19.9	15.8	1	20
Liquid NH ₃	18.6	12.7	1 or 10	−34 or 20
Liquid H ₂	120.0	8.5	1	−253
Compressed H ₂	120.0	7.5	700	20

**Figure 2.** Life cycle phases of a marine engine.

presented in Sections 2 and 3. Results and discussion are presented in Section 4 with the conclusions of this work drawn in Section 5.

Life cycle assessment

This study is a comparative attributional LCA study and impact assessment, based on the IPCC and CML2001 impact assessment methods. Five environmental indicators such as GWP100, GWP20, AP, EP, POCP were calculated as these indicators are mostly concerned by maritime industry regulations.³⁷ Calculations and results were obtained from LCA for Experts (GaBi) software application and database. Procedure to obtain the results from are following.

- (1) Finding the necessary processes/flows for the life cycle of the marine from the database (ecoinvent, Sphera).
- (2) Connecting processes to establish life cycle model.

- (3) Running model. GaBi has tools and many useful functions (including sensitivity analysis) which helps LCA practitioners to get results easily.

Goal definition

As presented above, the reason for carrying out this study is to compare the environmental impacts of H₂ and NH₃ with the usage of fossil fuels. The applications are to provide information to the research community and assist decision-making process in selecting the alternative fuels for decarbonization purpose. Maritime stakeholders, naval architects, ship-owners, policy makers, and the public are the intended audience of the study.

Figure 2 illustrates the diagram of system boundary and life cycle phases of marine engines. The life cycle of marine engine consists of the phases of material and fuel production, the operation and maintenance phase, production phase and end-of-life phase. In this study, the maintenance activities and human factor are

Table 2. Scenarios in this study.

Scenarios	Items	Unit	HFO/MDO	NH ₃	H ₂
2S_fossil	Energy contribution	-	100%	0%	0%
	Energy	kWh	1	0	0
	Mass	g	180.2	0	0
2S_GNH ₃	Energy contribution	-	5%	95%	0%
	Energy	kWh	0.05	0.95	0
2S_BNH ₃	Mass	g	9.01	392.08	0
	Energy contribution	-	100%	0%	0%
4S_fossil	Energy	kWh	1	0	0
	Mass	g	219.3	0	0
	Energy contribution	-	1.5%	0%	98.5%
4S_GH ₂	Energy	kWh	0.015	0	0.985
	Mass	g	3.3	0	76.7
4S_GNH ₃	Energy contribution	-	12%	88%	0%
	Energy	kWh	0.12	0.88	0
4S_BNH ₃	Mass	g	26.3	44.2	0

ignored in order to simplify. The inputs and outputs of the life cycle are energy, materials, and emissions flows.

Functional unit definition

The investigated product in this study is a main engine of ship, of which, the main function is to create energy in combustion process to propulsive the ship. *The functional unit chosen is 1 kW hour (1 kWh) delivered to the propeller shaft.* The results will be presented per one kWh, for example, kilogram of CO₂ eq./kWh.

Eight scenarios

In this study, eight scenarios with different marine fuels and engines were investigated (Table 2). 2S_fossil and 4S_fossil scenarios were chosen as the base case scenarios with 100% of fossil fuels used on ship. The scenarios and the fuel consumption of the ships were based on the real ship operation using chemical tankers (Stolt Tankers B.V.). Some difference exists in the size and age of the vessels and engines, given practical limitations of data and variations in vessel and engine size in the fleet. The 4-stroke powered ship consisted of a 37,059 DWT Chemical & Oil Carrier built in 2001 using a 9-cylinder 4-stroke engine with 320 mm bore and 350 mm stroke. It had a maximum continuous power rating of 10,944 kW. In the H₂ scenarios, H₂ (only in 4-stroke engines) were used as main fuel with 1.5% MDO necessary (percentage fraction given by energy content) as pilot injection along the balance of H₂ main fuel (98.5%). NH₃ scenarios used 88% NH₃ as main fuel. The two-stroke engine ship consisted of a 33,723 DWT Chemical & Oil Carrier, built in 2017 using a 5-cylinder engine with 500 mm bore, 2500 mm stroke, and a maximum continuous power rating of 5850 kW. The fraction of pilot fuel injection was assumed to be 5% MDO with 95% NH₃ as main fuel.

The fossil fuel and NH₃ scenarios were updated with the inclusion of SCR technologies in order to meet the

requirements in the NO_x emission limits of Regulation 13 of MARPOL Annex VI. The estimation will be 15 g of urea per 1 kWh could reduce 90% NO_x.³⁸ The application of SCR affected the results of NO_x emissions as well as the GWP in the LCA of marine engines. Since the interest towards NH₃ and H₂ as competitive candidates to reduce carbon emissions from shipping only gain larger attention in recent years, there are not yet consistent emission data available. Within the coming years new insight to the emission levels from NH₃ and H₂ with different engine technologies will very likely be published.

Assumptions and limitations

Assumptions and limitations are inevitable in LCA studies due to the numerous datasets and information in the product's life cycle. In this study, assumptions and limitations are listed as below:

1. Only steel was considered as the material used to produce the vessels' main engines, the secondary materials such as copper, aluminium, etc. were ignored.
2. The transportation mode used for transporting fuel from fuel production site (Porsgrunn, Norway) to the port (Port of Amsterdam, the Netherlands) was ocean-going vessels with the distance of 800 km.
3. The vessels and main engines will be retired and dismantled after 25 years operating at sea. In this study, end-of-life of the engine is considered.
4. Waste management and generation during the engine's life cycle were not considered.

Primary emissions

Most emissions from the shipping industry are released as exhaust gases and these emissions have impacts on the environment, leading to effects on the climate, human health and the marine environment among

Table 3. Sensitivity analysis description.

Scenarios	Sensitivity analysis (SA)							
	Emission factors of N ₂ O in engine combustion		MDO for pilot injection (%)		Rate of H ₂ boil-off (%)		Energy for H ₂ liquefaction (kWh/kgH ₂)	
	Upper limit	Lower limit	Upper limit	Lower limit	Upper limit	Lower limit	Upper limit	Lower limit
2S_fossil	-	-	-	-	-	-	-	-
2S_GNH ₃	10 times higher	50% of base case	10.0%	2.5%	-	-	-	-
2S_BNH ₃	-	-	-	-	-	-	-	-
4S_fossil	-	-	-	-	-	-	-	-
4S_GH ₂	-	-	3%	0%	1%	0.10%	10	6
4S_BH ₂	-	-	-	-	-	-	-	-
4S_GNH ₃	10 times higher	50% of base case	18%	6%	-	-	-	-
4S_BNH ₃	-	-	-	-	-	-	-	-

other. Since emissions to air are the most important emissions the foreground system is limited to the emissions to air. The list of primary emissions includes CO₂, PM, CH₄, NO_x and N₂O. Most of these emissions were investigated in the GHG report of the IMO.³⁹

Sensitivity analysis description

In order to validate the robustness of the results, sensitivity analyses were performed (Table 3). First, the rate of MDO used for pilot injection process was analysed for the H₂ and NH₃ scenarios. Second, the percentage of H₂ boil-off, which depends on the length of vessel's voyages and the boil-off H₂ rate (from 0.1% to 1% per day), was considered.⁴⁰ In the initial stage, 0.5% boil-off H₂ per day was used. The H₂ boil-off can lead to the extra fuel consumption due to the increase of energy used for re-liquefaction. Third, CO₂ capture rate in CCS technology (from 85% to 95%) and the energy demand for H₂ liquefaction (from 6 to 10 kWh/kgH₂)⁴¹ were also taken into account in the sensitivity analysis. Moreover, the N₂O emission factor in engine combustion were considered in this part.

Life cycle inventory

The life cycle inventory consists of the mass, energy and emission information for all inputs and outputs of life cycle system for the fossil fuels, H₂ and NH₃ engines. The data for this part were provided by GaBi software and data providers. Data were also collected and gathered from literature such as publications, technical reports, etc.

Fuel production and transportation

Fuel production database is available in GaBi software and database. After production phase, fossil fuels was transported by ocean-going vessels with the distance of 800km to the refuelling station. H₂ and NH₃ were assumed to be produced in Porsgrunn, Norway. These

two types of fuels were assumed to be transported by sea going vessel with the same distance with fossil fuels to the refuelling station.

In this work, the production of 'green' H₂ used the electricity from wind energy in Norway. In order to produce 1 kg of gaseous H₂ by electrolysis method, 192 MJ of electricity are consumed. After that, 36 MJ of electricity were assumed to be used for H₂ liquefaction process (1 kg H₂).⁴²

Steam reforming methane (or natural gas) is used to produce blue H₂. The carbon capture and storage in this process consumes 42.1 MJ/1 kg of H₂.⁴³ In this study, we assumed that 90% of carbon dioxide are captured in the H₂ production.

'Blue' H₂ was assumed to be used to produce 'blue' NH₃ by applying Haber-Bosch process. 0.824 kg of nitrogen from air separation, 0.176 kg of H₂ and 1.17 MJ of electricity⁴⁴ are consumed in order to produce 1 kg of NH₃.⁴⁵ The liquefaction process of NH₃ consumes less electricity than H₂, only 3.01 MJ/kg NH₃ (estimated by the authors).

The final production step in the production of 'green' NH₃ is similar to that of 'blue' NH₃. The 'green' H₂ and energy are used with the same assumption as 'blue' NH₃ production for the Haber-Bosch synthesis step.

Fuel combustion and ship operation

The fuel consumption was obtained from operational data for the year 2021 (Table 4). The average fuel consumption over 1 year was 180.2 g/kWh for the two-stroke engine case, and 219 g/kWh for the 4-stroke engine case, based on the vessel data.

Emission factors in ship operation are gathered from IMO's GHG report³⁹ and estimated from some publications (Table 5). As H₂ and NH₃ are carbon-free fuels, using H₂ and NH₃ in this phase does not emit CO₂, CH₄, black carbon, and CO. In addition, using H₂ and NH₃ can eliminate PM₁₀, PM_{2.5}, SO_x and NMVOC emissions. However, a small amount of N₂O and NO_x

Table 4. Fuel consumption of vessels in 2021 from noon report.

Engines	2-stroke engine vessel	4-stroke engine vessel
Average speed	14.5 knot	16.2 knot
Main engine power	5850 kW	10,944 kW
Auxiliary engines power	940 kW × 3	2430 kW × 2
Total fuel consumption	5054.5 ton HFO 758.8 ton MDO	5981.9 ton HFO 1648 ton MDO

is still generated due to the combustion of NH₃ and nitrogen. There are also some un-burned H₂ and NH₃ emitted to the air in this phase. However, the data for tank-to-wake for H₂ and NH₃ are very limited and it can be different. This will be considered in the sensitivity analysis of this study.

Material used in engine production

The cradle-to-grave life cycle of material was considered in this study. It included mainly the raw material extraction and production. After 25 years of operation, the engine will be scrapped and used for recycling.

As the chosen functional unit is one kWh delivered to the propeller shaft, the amount of material considered per each functional unit is defined by dividing the mass of engine by the total energy produced in the engine's life cycle (kg of steel per kWh). The secondary material such as aluminium, copper, etc. were excluded due to the small amount, compared to the mass of steel.

Results and discussion

Life cycle emissions

Figure 3 presents the contribution of life cycle phases to the total mass of CO₂ in the life cycle of marine engine per unit of energy of shaft power delivered. Generally,

CO₂ emissions could be effectively reduced by using H₂ and NH₃ solutions, compared to fossil fuels scenario. However, for 'blue' H₂ and 'blue' NH₃, CO₂ emissions from production phase (WTT) are much higher than for the WTT phase of fossil fuels.

The CH₄ emissions are emitted mostly from the production phase (Figure 4). It should be noted that, due to the use of grid-electricity in the production phase, the amount of these emissions for 'blue' solutions are much higher than the 'green' ones, even higher than the amount of CH₄ in fossil fuel production phase.

Figure 5 illustrates PM emissions from marine engine's life cycle. The use phase of fossil fuels emit the higher amount of PM than H₂ and NH₃ scenarios. Meanwhile, the H₂ and NH₃ scenarios could reduce effectively these emissions. Regarding black carbon, fossil fuel scenario also generated much more amount of these emissions.

The application of NH₃ in ship operation could increase the amount of N₂O emissions (Figure 6). NH₃ can only reduce a small amount of NO_x (Figure 7) and SCR needs to be used in the scenarios of fossil fuels and NH₃.

Environmental indicators

Figure 8 illustrates the GWP of eight investigated scenarios in this study. It is clear that fossil fuel scenarios have higher GWP value than H₂ and NH₃ solutions. More than 50% of GWP could be reduced by using green H₂ or NH₃ as marine fuels (Table 6). When using fossil fuels, most of GHG emissions are emitted from the tank-to-wake phase, meanwhile, the GHG emissions of H₂ and NH₃ depend on the production phases. For 4S_GNH3 and 4S_BNH3, the use of fossil fuels considerably increase the amount GWP.

Table 7 summarizes the environmental indicators results. Although the use of NH₃ as marine fuel could reduce the impact on climate change, it still has higher AP and EP values than fossil fuel scenarios.

Table 5. Emission factors in fuel combustion process (kg emission/kg fuel).

Emissions	MDO ³⁹	HFO ³⁴	H ₂	NH ₃
CO ₂	3.20600	3.114	0.00000	0.00000
CH ₄	0.00001	0.00005	0.00000	0.00000
N ₂ O	0.00018	0.0759	0.00000	0.00033 ⁴⁶
NO _x	0.05671	0.00018	0.02333 ⁴⁷	0.02033 ⁴⁸
CO	0.00259	0.00288	0.00000	0.00000
NM VOC	0.00240	0.0032	0.00000	0.00000
SO _x	0.00137	9.7752E-06	0.00000	0.00000
PM10	0.00090	0.00755	0.00000	0.00000
PM2.5	0.00083	0.00694	0.00000	0.00000
Black carbon (soot)	0.00038	0.00026	0.00000	0.00000
H ₂ (unburned)	0.00000	0.00000	0.00800 ^{49*}	0.00000
NH ₃ (unburned)	0.00000	0.00000	0.00000	0.00950 ⁴⁸

*Assumed 0.2% unburned fuel.

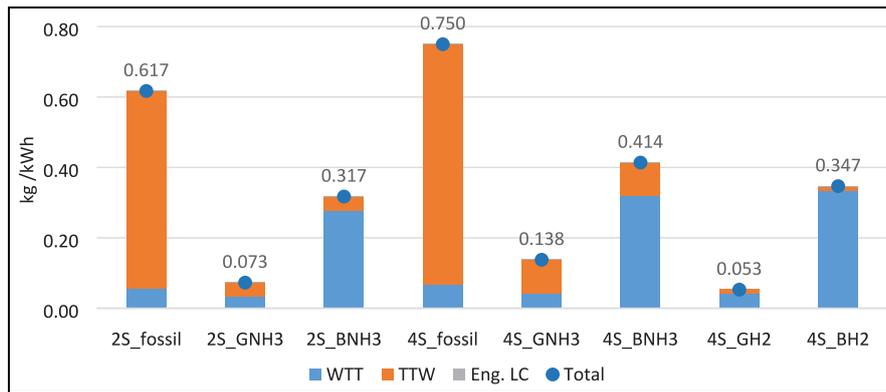


Figure 3. Life-cycle CO₂ emissions.

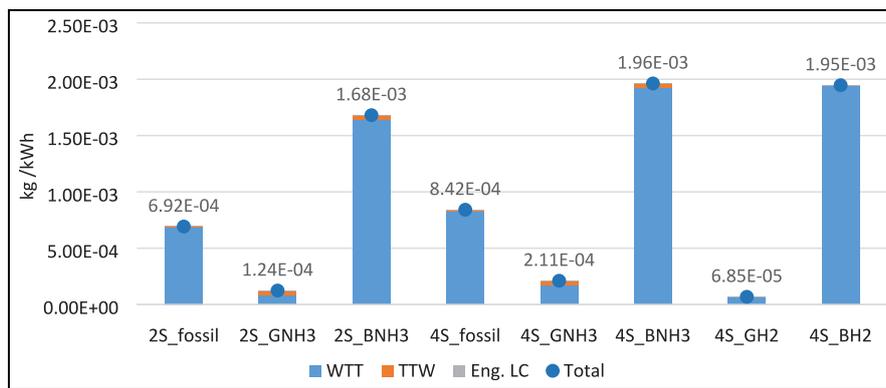


Figure 4. Life-cycle CH₄ emissions.

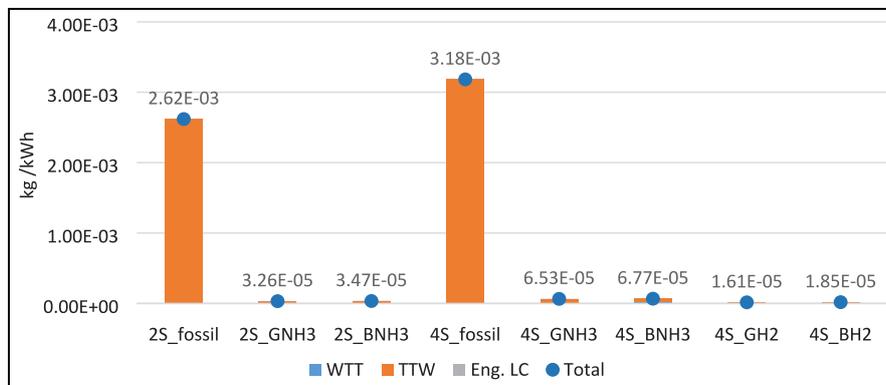


Figure 5. Life cycle PM emissions.

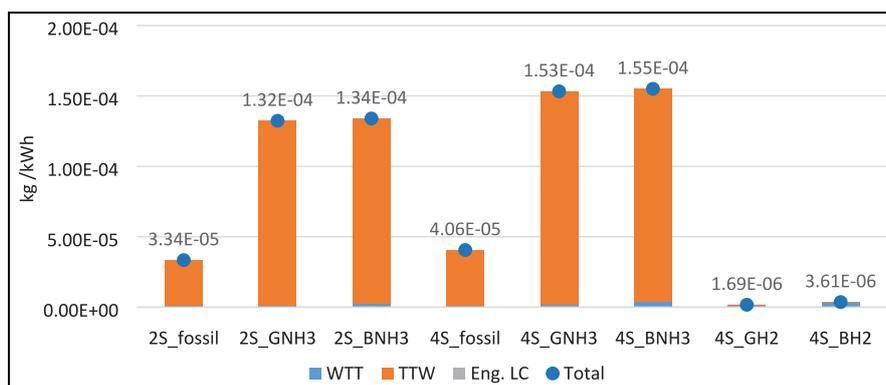


Figure 6. Life-cycle N₂O emissions.

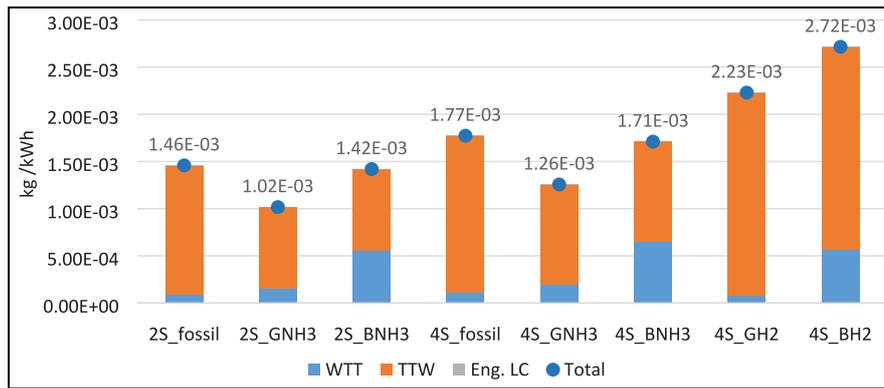


Figure 7. Life cycle NO_x emissions.

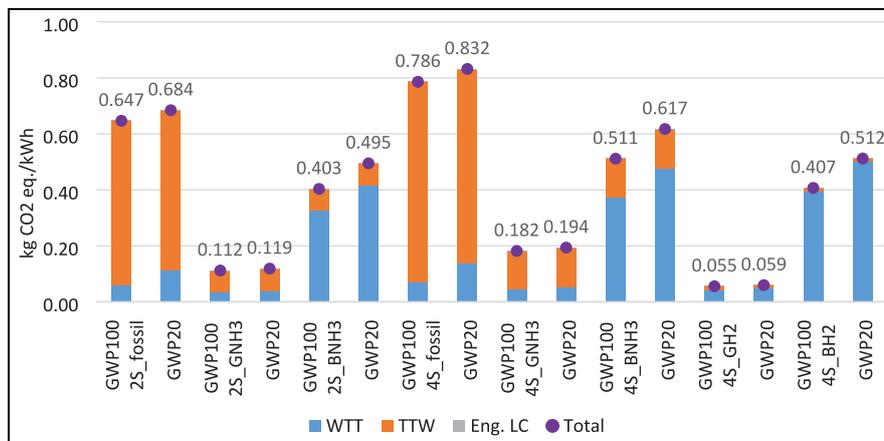


Figure 8. GWP results.

Table 6. GWP comparison between eight scenarios (unit: kg CO₂ eq./kWh).

Scenarios	GWPI00	Comparisons (%)	GWP20	Comparisons (%)
2S_fossil	0.647	100.00	0.684	100.00
2S_GNH3	0.112	17.31	0.119	17.40
2S_BNH3	0.403	62.29	0.495	72.37
4S_fossil	0.786	100.00	0.832	100.00
4S_GNH3	0.182	23.16	0.194	23.32
4S_BNH3	0.511	65.01	0.617	74.16
4S_GH2	0.055	7.00	0.059	7.09
4S_BH2	0.407	51.78	0.512	61.54

Table 7. Environmental indicators.

Scenarios	GWPI00	GWP20	AP	EP	POCP
2S_fossil	6.47E-01	6.84E-01	1.15E-03	2.11E-04	6.38E-14
2S_GNH3	1.12E-01	1.19E-01	6.67E-03	1.48E-03	5.04E-14
2S_BNH3	4.03E-01	4.95E-01	6.81E-03	1.54E-03	1.30E-13
4S_fossil	7.86E-01	8.32E-01	1.40E-03	2.54E-04	7.37E-14
4S_GNH3	1.82E-01	1.94E-01	7.62E-03	1.69E-03	5.86E-14
4S_BNH3	5.11E-01	6.17E-01	7.77E-03	1.75E-03	1.49E-13
4S_GH2	5.55E-02	5.94E-02	1.31E-03	2.97E-04	7.19E-14
4S_BH2	4.07E-01	5.12E-01	1.48E-03	3.60E-04	1.65E-13

Units: GWPI00 (kg CO₂ eq./kWh), GWP20 (kg CO₂ eq./kWh), AP (kg SOX eq./kWh), EP (kg phosphate eq./kWh), ODP (kg R11 eq./kWh), POCP (kg ethene eq./kWh).

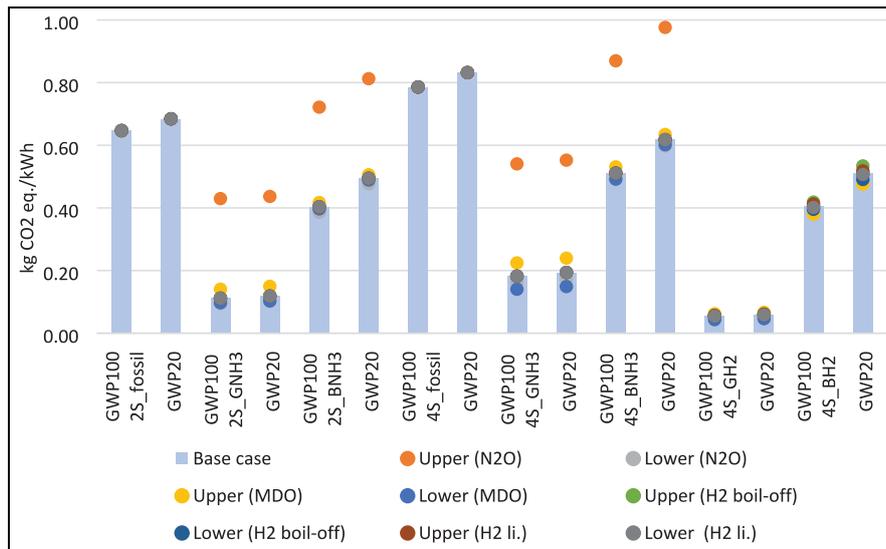


Figure 9. Sensitivity analysis of GWP results.

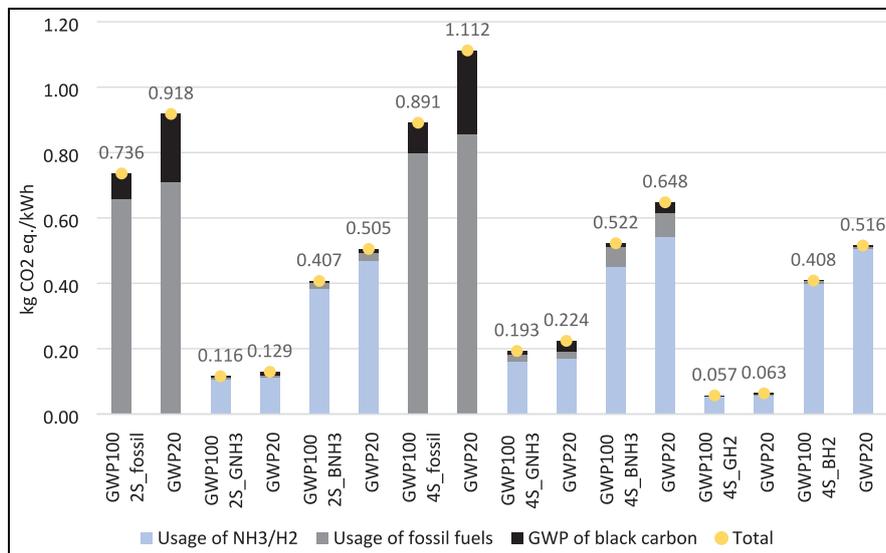


Figure 10. Black carbon effect.

Sensitivity analysis and black carbon

Figure 9 shows the sensitivity analysis of the GWP results. It can be seen that the most influential factor on the GWP results is N₂O emission factor for NH₃ scenarios. The upper N₂O emission factor can lead to 70.1% increase in GWP100 in 4S_BNH3 scenario, which is higher than GWP100 of 4S_fossil scenario. Therefore, blue NH₃ should be carefully considered when using as maritime fuel and the N₂O emission factor is needed to be investigated to support the decision-making process. The results for NH₃ scenarios are also greatly affected by the rate of pilot fuel, for example, 10% increase in rate of MDO leads to 23.1% increase in GWP20 value for 4S_GNH3 scenario. The green H₂ and NH₃ still show the lower value of GWP than fossil fuel scenarios.

In case the characterization factors of black carbon for GWP100 and GWP20 are 1647.5 and 4470⁵⁰ respectively, the contribution of black carbon to the GWP results can be seen in Figure 10. The use of fossil fuels clearly bring the considerable impact of black carbon to GWP, especially to GWP20. Therefore, fossil fuel should be cut down in order to achieve the IMO’s decarbonization goals in 2030 and 2050. The rate and type of pilot fuel should be also considered and it depends on engine technologies in the future.

The advantage of using H₂ fuel is the lower GWP value, compared with NH₃. H₂ fuel also cuts down the amount of black carbon due to the small percentage of fossil fuels used in engines as pilot fuel. However, H₂ fuel is now limited used in 4-stroke engine. The use of H₂ on the ships that have long voyage should be

carefully considered because of the boil-off rate of H₂. These ships require considerable amount of energy to keep H₂ on board and it could lead to the increase of GWP value of marine engines in the life cycle perspective.

Conclusions

The LCA methodology has been applied in this work in order to investigate the environmental performance of marine engines associated with eight scenarios using fossil fuels, 'blue' and 'green' H₂ and NH₃. In the LCA, the production of engine, fuel (well-to-tank), the usage phase (tank-to-wake) and the engine's end-of-life were examined.

The initial results show that NH₃ and H₂ solutions could reduce the environmental impacts, in comparison with fossil fuels. The 'green' fuels are more environmentally friendly than the 'blue' ones. However, it should be noted that, the production of NH₃ and H₂ phases dominate the life cycle results. Different fuel production regions will bring different results in environmental performance of marine engines.

Due to the boil-off of H₂ in the storage H₂ on-board, penalty energy is required for the re-liquefaction process. Therefore, it seems that H₂ fuel does not show its advantages for the longer voyage. The space used for H₂ storage is also higher than NH₃ case. Generally, H₂ solution has less impact on the environment than the NH₃ solution in the system boundary of our work.

The limitation of this study is that it does not consider economic and social aspects of hydrogen/ammonia engines. The results of this study represent the current state-of-the-art technology, and future technologies might bring further improvements in not only environmental aspects but also societal and economic performance.

To meet IMO's ambitious GHG targets, it is clear that green NH₃ and H₂ are potential candidates for cutting down the emissions from the shipping industry. However, some significant issues that are required attention such as fuel infrastructure, marine fuel logistics, cost benefits, safety aspects, etc. This ensures the advantages of alternative marine fuel application towards the sustainable shipping industry in the future.

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Appendix

Abbreviation

2S_BNH ₃	two-stroke engine using blue ammonia	GWP ₂₀	global warming potential (time horizon: 20 years)
2S_GNH ₃	two-stroke engine using green ammonia	H ₂	hydrogen
2S_fossil	two-stroke engine using fossil fuels	HFO	heavy fuel oil
4S_BH ₂	four-stroke engine using blue hydrogen	IMO	International Maritime Organization
4S_BNH ₃	four-stroke engine using blue ammonia	IPCC	Intergovernmental Panel on Climate Change
4S_GH ₂	four-stroke engine using green hydrogen	LCA	life cycle assessment
4S_GNH ₃	four-stroke engine using green ammonia	LNG	liquefied natural gas
4S_fossil	four-stroke engine using fossil fuels	MDO	marine diesel oil
AP	acidification potential	HFO	heavy fuel oil
CCS	carbon capture & storage	N ₂	nitrogen
CH ₄	methane	N ₂ O	nitrous oxide (laughing gas)
CO	carbon monoxide	NH ₃	ammonia
CO ₂	carbon dioxide	NMVOOC	non-methane volatile organic compounds
Eng. LC	life cycle of material used to produce engines	NO _x	nitrogen oxides
EP	eutrophication potential	PM	particulate matter
eq.	equivalent	POCP	photochemical ozone creation potential
GWP ₁₀₀	global warming potential (time horizon: 100 years)	SCR	selective catalytic reduction
		TTW	tank-to-wake
		VOCs	volatile organic compounds
		WTT	well-to-tank
		WTW	well-to-wake